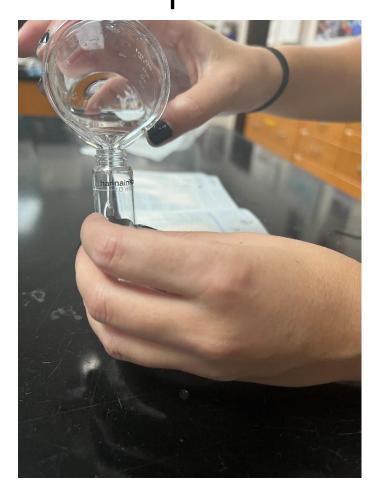
How to use the **Phosphate**Hanna Meter

Step 1: Press (tap) the on/off button to turn the checker on

• Note: All segments will be displayed for a few seconds, followed by "ADD", "C1", with "Press" blinking

Step 2: Fill the cuvette with 10 ml of sample and replace the cap

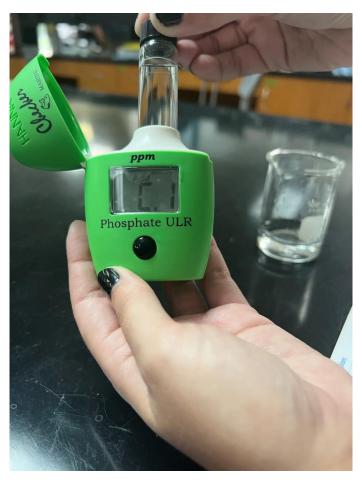




Step 3: Wipe off the cuvette. Make sure there are no droplets on the outside of the cuvette

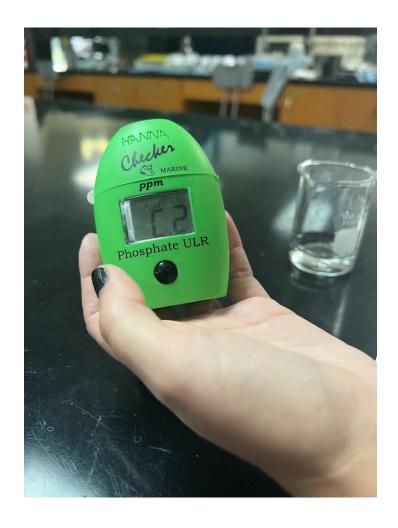


Step 4: Place cuvette into the checker, close cap



Step 5: **Press (tap)** the on/off button. When the display shows "ADD", "C.2", with "Press" blinking, the checker is zeroed





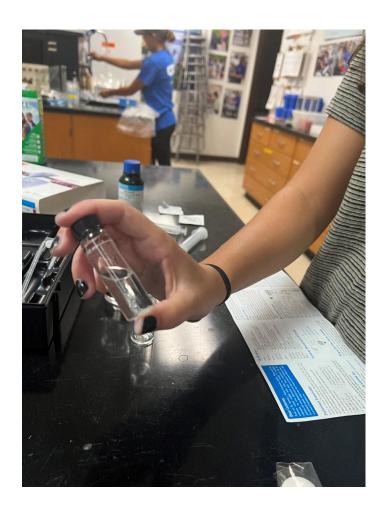
Step 6: Remove the cuvette, unscrew the cap and add the content of one packet of **HI774-0**





Step 7: Replace the cap and shake GENTLY for about 2 minutes until the powder is completely

dissolved.



Step 8: Insert the cuvette back into the checker and close cap

Note: Please make sure there is no droplets on the outside of the

cuvette before placing in the checker



Step 9: **Press** and **HOLD** (for about 3 seconds) the on/off button. Release when the display shows a countdown. This countdown represents the checker

reading the sample.



Step 10: When the timer ends the checker will perform the reading

• **Note:** The instrument displays the phosphate concentration in ppm. The checker automatically turns off 2 minutes after reading

Phosphorus is one of the primary nutrients that regulates the growth of algae and larger aquatic plants, particularly in fresh water. Phosphate, the form in which almost all phosphorus is found in water, can enter the aquatic environment in a number of ways. Natural processes transport phosphate to water through atmospheric deposition, ground water percolation, and terrestrial runoff. Municipal treatment plants, industries, agriculture, and domestic activities also contribute to phosphate loading through direct discharge and natural transport mechanisms. The very high levels of phosphorus in some of Florida's streams and estuaries are usually caused by phosphate mining and fertilizer processing activities. High phosphorus concentrations are frequently responsible for accelerating the process of eutrophication (or accelerated aging) of a waterbody. Once phosphorus and other important nutrients enter the ecosystem, they are extremely difficult to remove because they are taken up by plants or deposited in sediments. Nutrients, particularly phosphates, deposited in sediments generally are redistributed into the water. This type of cycling compounds the difficulty of halting the eutrophication process.

Measured in mg/l

Percentile	Blackwater	Coastal	Estuary
10	0.02	0.01	0.01
20	0.04	0.01	0.02
30	0.06	0.01	0.05
40	0.09	0.01	0.07
50	0.11	0.02	0.10
60	0.14	0.03	0.12
70	0.18	0.04	0.16
80	0.24	0.06	0.22
90	0.38	0.12	0.33

Typical Values for Water Quality Parameters in the State of Florida

1mg/l = 1ppm A reading of 0.1ppm is a median reading in a Florida estuary.

Water Quality Test Error Code Cheat Sheet:

Error Code	Meaning	
L.Hi	Too much light hitting detector, check	
	preparation of zero cuvette.	
L.Lo	Not enough light hitting detector, check	
	preparation of zero cuvette.	
Inu	Sample and zero cuvettes inverted. Swap and	
	repeat measurements.	
0.00 (blinking)	Under range (sample absorbed less light than	
	zero).	
0.90 (blinking)	Over range (Measured value outside limits of	
	checker device). Make sure there is no debris in	
	sample and dilute/repeat.	
bAt	Replace battery.	
bAd	Replace battery.	

Indian River Lagoon Water Quality Parameter Cheat Sheet:

Water Quality Test	Average Range in IRL	Unit
Temperature	75-85	Degrees Fahrenheit
PH	6-8	No unit
Salinity	15-25	PPT (parts per thousand)
Nitrates	0.01-0.10	PPM (parts per million)
Nitrites	0.00-0.02	PPB (parts per billion)
Phosphates	0.02-0.22	PPM (parts per million)
Alkalinity	66-133	PPM (parts per million)
Turbidity	0-100 JTU	JTU (Jackson Turbidity Unit) if using LaMotte kit or cm
	0-450 cm	(centimeters) if using secchi disk- probes may have different units
DO	0-5	PPM (parts per million)